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Triterpene Glycosides from Sea Cucumber *Holothuria scabra* with Cytotoxic Activity

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Abstract: **Objective** To study the new triterpene glycosides from sea cucumber *Holothuria scabra* with cytotoxic activity.**Methods** Triterpene glycosides from *H. scabra* were separated and purified by chromatography on DA-101, silica gel, and reversed-phase silica gel column, as well as RP-HPLC. Their structures were elucidated on the basis of spectral data and chemical evidence. **Results** Three triterpene glycosides were identified as scabraside D (**1**), fuscocineroside C (**2**), and 24-dehydroechinoside A (**3**). Their inhibition on P-388, A549, MKN-28, HCT116, and MCF-7 cells were significant. **Conclusion** Scabraside D (**1**) is a new triterpene glycoside, and compounds **2** and **3** are isolated from *H. scabra* for the first time. The glycosides **1–3** show the *in vitro* cytotoxicity against five human tumor cell lines in comparison to 10-hydroxycamptothecin.**Key words:** cytotoxicity; 24-dehydroechinoside A; fuscocineroside C; *Holothuria scabra*; scabraside D; triterpene glycoside**DOI:** 10.3969/j.issn.1674-6384.2012.03.002

Introduction

Sea cucumbers, *Holothuria scabra* Jaeger (Holothuriidae), are widely distributed in Atlantic and Pacific Oceans and have been used in traditional Chinese medicine as tonics and delicacies for a long time. Triterpene glycosides are the most important secondary metabolites in sea cucumbers because of their biological activities, including antifungal, cytotoxic, hemolytic, cytostatic, and immunomodulatory effects (Habermehl and Volkvein, 1971; Kitagawa *et al.*, 1989; Stonik, Kalinin, and Avilov, 1999; Chludil *et al.*, 2002; Zhang, 2011a; 2011b). More than 100 compounds have been isolated from sea cucumbers up to date. Most of the known sea cucumber glycosides have lanostane aglycones with an 18(20)-lactone and a sugar chain composed of up to six monosaccharide units linked to the C-3 of the aglycone, which is

composed of *D*-xylose, *D*-quinovose, *D*-glucose, and 3-*O*-methyl-*D*-glucose (Stonik and Elyakov, 1988; Maier *et al.*, 2001). Sea cucumber is abundantly distributed in the South China Sea. Some triterpene glycosides have been isolated from *H. scabra* (Liao, 1997; Han *et al.*, 2009a; 2009b). In a preceding paper, we also reported the antifungal and cytotoxic activities of this sea cucumber. As a part of our research on biological secondary metabolites from echinoderms (Zou *et al.*, 2003; Tang *et al.*, 2005; Zhang, Yi, and Tang, 2006; Yi *et al.*, 2006; Han *et al.*, 2007; Sun *et al.*, 2007) and a continuation of studies on this sea cucumber, we present here the isolation and structure elucidation of a new sulfated triterpenoid glycoside, named scabraside D (**1**), as well as the cytotoxicities of the three glycosides against five human tumor cell lines.

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Materials and methods

Equipments

Melting points were determined on an XT5-XMT apparatus. Optical rotations were measured with a Perkin-Elmer 341 Polarimeter. IR spectra were recorded on a Bruker Vector 22 Infrared Spectrometer. NMR spectra were recorded in C_5D_5N on a Varian Inova-600 Spectrometer, and the 2D NMR spectra were obtained using standard pulse sequences. ESI-MS and HR-ESI-MS were recorded on a Micromass Quattro Mass Spectrometer. GC-MS was performed on a Finnigan Voyager Apparatus using a DB-5 column (30 m \times 0.25 mm, 0.25 μ m) with an initial temperature of 150 $^{\circ}C$ for 2 min and then temperature programming to 300 $^{\circ}C$ at a rate of 15 $^{\circ}C$ /min. Semi-preparative HPLC was carried out on an Agilent 1100 Liquid Chromatograph equipped with a Refractive Index Detector using a Zorbax 300 SB- C_{18} column (25 cm \times 9.4 mm). Column chromatographies were performed on silica gel (200–300 meshes, 10–40 μ m; China) and ODS (40–63 μ m; Merck, Germany) and Sephadex LH-20 (Pharmacia). Fractions were monitored by thin layer chromatography (TLC) [precoated silica gel GF₂₅₄ plates (10–40 μ m; China)], and spots were visualized by heating silica gel plates sprayed with 15% H_2SO_4 in EtOH.

Experimental materials

Specimens of *Holothuria scabra* Jaeger were collected from offshore water of Hainan Island in the South China Sea in May, 2006, and authenticated by Prof. LIAO Yu-lin (Institute of Oceanology, Chinese Academy of Sciences, China). A voucher specimen (HY200605) was deposited at the Research Center for Marine Drugs, School of Pharmacy, Second Military Medical University.

Extraction and isolation

The sea cucumbers (3 kg, dry weight) were powdered and refluxed for four times with 60% ethanol (6 L \times 4, each time for 1 h). The extract was concentrated, and the residue (420 g) was suspended in H_2O , passed through a DA101 resin column (2 kg, 105 cm \times 15 cm, Nankai University, China) and then eluted with H_2O (5 L), 70% EtOH (10 L), and 95% EtOH (5 L), respectively. The glycoside fraction was eluted with 70% ethanol. The combined extracts were

concentrated. The glycoside fraction (crude glycoside-containing mixture, 70 g) was separated over silica gel column chromatography (CC, 200–300 meshes, 2.1 kg), stepwise eluted with $CHCl_3$ -MeOH- H_2O (8:2:1 to 6.5:3.5:1, lower phase) gradient to give Frs. A (2.43 g), B (3 g), C (1.13 g), D (3.8 g), and E (2.23 g).

Fr. E was subjected to CC (ODS RP- C_{18} ; MeOH- H_2O 54:46) and gave subfractions E_1 and E_2 . Subfraction E_1 was purified by HPLC (Zorbax 300 SB- C_{18} ; 59% MeOH, 1.5 mL/min) to afford compounds **1** (36.6 mg; t_R = 20.1 min) and **2** (11 mg; t_R = 26.7 min). Subfraction E_2 gave 110 mg of pure glycoside **3** (t_R = 29.3 min) using MeOH- H_2O (62:38) as the mobile phase and a flow rate of 1.5 mL/min.

Acid hydrolysis of compounds 1–3

Each of the glycoside (1 mg) was heated with 2 mol/L trifluoroacetic acid (1 mL) at 120 $^{\circ}C$ for 2 h. The reaction mixture was evaporated to dryness and the residue was partitioned between CH_2Cl_2 and H_2O . The aqueous phase was concentrated under reduced pressure. Then pyridine (1 mL) and $NH_2OH \cdot HCl$ (2 mg) were added to the dried residue, and the mixture was heated at 90 $^{\circ}C$ for 30 min. Then, Ac_2O (0.8 mL) was added, and heating was continued at 90 $^{\circ}C$ for 1 h. The solution was concentrated and the resulting aldononitrile peracetates were analyzed by GC-MS using standard aldononitrile peracetates as reference samples. *D*-xylose (Xyl), *D*-quinovose (Qui), *D*-glucose (Glc), and *D*-3-*O*-methylglucose (MeGlc) were identified in a 1:1:1:1 ratio for all the glycosides (*D*-Xyl: t_R = 5.53 min; *D*-Qui: t_R = 5.44 min; *D*-Glu: t_R = 6.75 min; *D*-3-*O*-MeGlc: t_R = 6.57 min).

Bioassay

The cytotoxicities of compounds **1–3** (95% purity) against mouse leukemic cell (P-388), human lung cancer cell (A-549), gastric cancer cell (MKN-28), human colorectal cancer cell (HCT-116) and human breast cancer cell (MCF-7) (Shanghai Institute of Materia Medica, Chinese Academy of Sciences) were evaluated by sulforhodamine-B (SRB) assay (Skehan *et al.*, 1990), with the anticancer agent 10-hydroxycamptothecin (HCP, 98% purity; Knowshine Pharmaceuticals Inc.; China) as a positive control. IC_{50} was determined graphically for each experiment by curve fitting using Prism 4.0 software (GraphPad software, Inc.) and the equation derived by DeLean,

Munson, and Rodbard (1978). The results showed that the compounds exhibited significant cytotoxicity against the five tumor cell lines.

Results and discussion

Structure elucidation

Scabraside D (**1**) was positive in the Libermann-Burchard and Molish tests. Its molecular formula was determined as $C_{54}H_{87}O_{27}SNa$ from pseudomolecular ion peak at m/z 1245.5864 $[M + Na]^+$ (calcd. for $C_{54}H_{87}O_{27}S^+Na_2$: 1245.5860) in positive-ion mode HR-ESI-MS and at m/z 1199 $[M - Na]^-$ in negative-ion mode ESI-MS. A fragment ion peak at m/z 1125 $[M - OSO_3Na + Na - H]^+$ indicated the presence of a sulfate group in compound **1**, which was confirmed by the IR spectrum with absorption bands at 1266 and 1074 cm^{-1} . An examination of 1H -NMR and ^{13}C -NMR spectra of compound **1** indicated the presence of a triterpene aglycone with seven methyls, one olefinic bond and one lactone carbonyl group, which had a close similarity to the aglycone of echinoside A (Kitagawa *et al.*, 1985) and holothurin A₃ (Dang *et al.*, 2007), but compound **1** differed from echinoside A at C-25 and holothurin A₃ at C-22. The position of a hydroxyl group at C-25 was deduced from the NMR signals at δ 81.4 (C-25), 38.5 (C-24), and δ 1.62 (m, H-24) together with the analyses of TOCSY and HMBC experiments. The HMBC spectrum showed cross-peaks H-27/C-25, H-24/C-25, and H-26/C-25. The double bond at $\Delta^{9(11)}$ was deduced from the NMR signals at δ 153.8 (C-9), 115.6 (C-11); and δ 5.60 (br d, $J = 10.4$ Hz, H-11) together with the analyses of TOCSY and HMBC experiments. The HMBC spectrum showed cross-peaks H-9/C-11, H-19/C-9, H-8/C-11, and H-12/C-11, and in the TOCSY spectrum, two protons [δ 5.60 (H-11) and 4.95 (H-12)] comprised a two-spin system. A signal characteristic for an oxygenated methine [δ 71.6 (C-12) and 4.95 (m, H-12)] in the holostane nucleus indicated α -configuration of the allylic OH group at C-12 (Silchenko *et al.*, 2005). Therefore, a 12-hydroxylated $\Delta^{9(11)}$ terpenoid aglycone was identified.

The presence of four β -sugar units in compound **1** was deduced from the ^{13}C -NMR and 1H -NMR spectra, which showed four anomeric carbons and four anomeric protons resonances with coupling constant doublets ($J = 7.2$ – 8.0 Hz). The sugar moieties were

confirmed to be *D*-Xyl, *D*-Qui, *D*-Glc, and 3-*O*-methylglucose (MeGlc) at a ratio of 1:1:1:1 by acidic hydrolysis (2 mol/L CF_3COOH) followed by GC-MS analysis of the corresponding aldonitrile peracetates and by comparing the GC retention time of the corresponding aldonitrile peracetates with those of the authentic samples prepared in the same manner (Silchenko *et al.*, 2005). The 1H -NMR and ^{13}C -NMR signals attributable to the sugar units were assigned by the 2D NMR experiments and the data indicated that sugar residues were all in pyranose form. The sequence of the sugar residues in compound **1** was determined by analysis of HMBC correlations: Xyl H-1/C-3 of the aglycone, Qui H-1/Xyl C-2, Glc H-1/Qui C-4, and MeGlc H-1/Glc C-3. The position of the sulfate group was determined by comparing ^{13}C -NMR data of compound **1** with those of known glycosides (Breitmaier and Voelter, 1987). A downfield esterification shift was observed for the signal of Xyl C-4 (from δ 68.2 to 75.7). On the basis of the above data, the structure of compound **1** was deduced as 3-*O*-[3-*O*-methyl- β -*D*-glucopyranosyl-(1 \rightarrow 3)- β -*D*-glucopyranosyl-(1 \rightarrow 4)- β -*D*-quinovo-pyranosyl-(1 \rightarrow 2)-4-*O*-sulfate- β -*D*-xylopyransyl]-holosta-9(11)-ene-3 β ,12 α ,17 α ,25 α -tetrol and named scabraside D (Figs. 1 and 2).

Cytotoxicity

Some triterpene glycosides hitherto isolated from sea cucumber exhibited cytotoxic activity. Glycosides **1**–**3** isolated from the sea cucumber *H. scabra* were tested for *in vitro* cytotoxicity against five tumor cell lines (P-388, A549, MKN-28, HCT116, and MCF-7). HCP was used as a positive control. The results in Table 1 indicated that three glycosides showed cytotoxic activity against five tumor cell lines with IC_{50} in the range of 0.93–2.60 μ mol/L. On the basis of the data available, the cytotoxic activity of sea cucumber glycosides is very sensitive to their precise functionalization, and perhaps they show different sensitivities against different cell lines. Therefore, more extensive studies are needed before a clear structure-activity relationship could be reached. Based on these promising preliminary results, glycosides **1**–**3** need further study to be potential anticancer agents.

Compound **1**: colorless amorphous powder; mp 268 – 270 $^{\circ}C$, $[\alpha]_D^{20}$ -12.4° (c 0.4, MeOH); IR ν_{max}^{KBr} (cm^{-1}): 3417, 1773, 1632, 1266, 1074; ESI-MS

(+) mode: m/z : 1245 $[M+Na]^+$, (–) mode: m/z : 1199 $[M-Na]^-$; HR-ESI-MS (+) mode: m/z : 1245.5864 $[M+Na]^+$ (calcd. for $C_{54}H_{87}O_{27}S^+Na_2$: 1245.5860). 1H -NMR and ^{13}C -NMR are in Table 2.

Compound **2**: white crystal. It gave a positive reaction to Libermann-Burchard and Molish tests. mp

254 – 255 °C, $[\alpha]_D^{20}$ -3.5° (c 0.82, pyridine); IR ν_{max}^{KBr} (cm^{-1}): 3433, 1756, 1673, 1264, 1217; ESI-MS (+) mode: m/z 1227 $[M+Na]^+$ ($C_{54}H_{85}O_{26}S^+Na_2$), ESI-MS (–) mode: m/z 1181 $[M-Na]^-$. It was deduced as fuscocineroside C by comparison of the NMR spectra with those of fuscocineroside C (Zhang, Yi, and

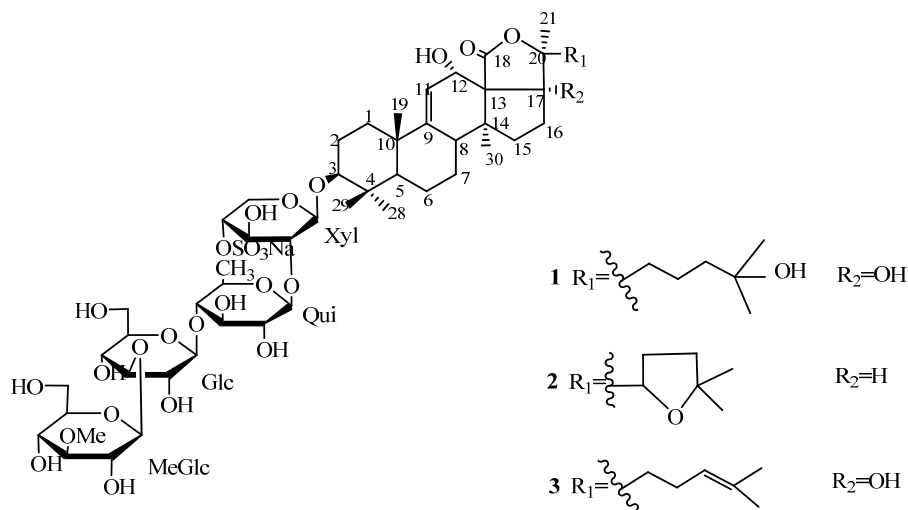


Fig. 1 Structures of compounds 1–3

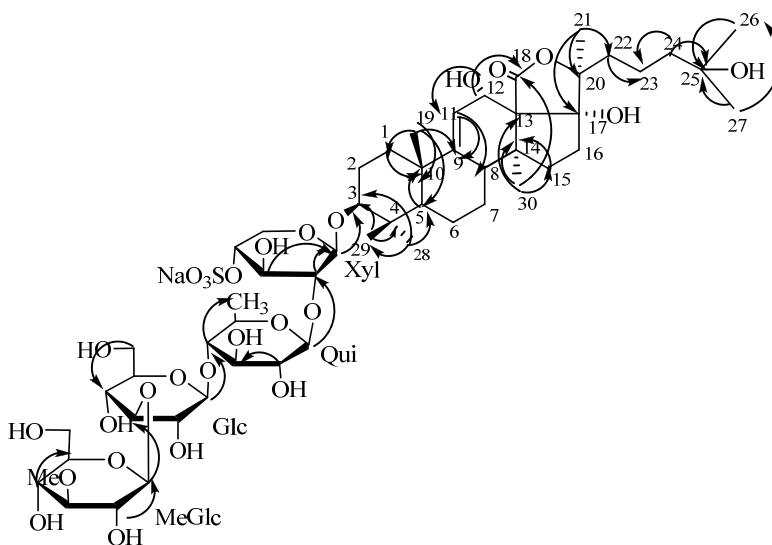


Fig. 2 Key HMBC correlations of compound 1

Table 1 Cytotoxicity of glycosides 1–3 against five tumor cell lines *in vitro* ($\bar{x} \pm s$, $n = 3$)

Cell lines	IC_{50} / ($\mu mol \cdot L^{-1}$)			
	1	2	3	HCP
P-388	0.96 ± 0.10	0.94 ± 0.11	1.23 ± 0.15	0.41 ± 0.13
A-549	1.72 ± 0.14	1.69 ± 0.12	$1.41 \pm 0.16^*$	0.84 ± 0.05
MKN-28	$1.27 \pm 0.13^*$	$0.93 \pm 0.10^*$	2.18 ± 0.32	0.77 ± 0.19
HCT-116	1.63 ± 0.17	1.70 ± 0.18	$1.11 \pm 0.13^*$	1.21 ± 0.14
MCF-7	$1.80 \pm 0.23^*$	2.60 ± 0.44	1.79 ± 0.21	1.17 ± 0.13

* $P < 0.05$ vs control

Table 2 ^1H -NMR (600 MHz) and ^{13}C -NMR (150 MHz) data for glycoside 1 in $\text{C}_5\text{D}_5\text{N}$

Position	δ_{H} (J)	δ_{C}	Position	δ_{H} (J)	δ_{C}	Position	δ_{H} (J)	δ_{C}
1	1.37 m, 1.82 m	36.4	21	1.74 s	18.9	4	3.64 m	86.8
2	1.89 m, 2.08 m	27.0	22	1.68 m	38.0	5	3.74 m	71.9
3	3.13 dd (4.2, 12.0 Hz)	88.7	23	2.03 m	28.4	6	1.70 d (6.0 Hz)	18.0
4		39.9	24	1.62 m	38.5	Glc		
5	0.98 m	52.8	25		81.4	1	4.96 d (7.8 Hz)	105.2
6	1.50 m, 1.76 m	21.2	26	1.19 s	28.7	2	4.09 m	75.4
7	1.72 m, 1.54 m	28.1	27	1.17 s	27.4	3	4.24 m	88.0
8	3.34 brd (10.2 Hz)	40.9	28	1.05 s	16.7	4	4.10 m	69.5
9		153.8	29	1.24 s	28.1	5	4.00 m	77.7
10		39.7	30	1.66 s	20.3	6	4.45 m, 4.42 m	61.8
11	5.60 brd (10.4 Hz)	115.6	Xyl			MeGlc		
12	4.95 m	71.6	1	4.66 d (7.2 Hz)	104.8	1	5.31 d (8.0 Hz)	105.8
13		58.9	2	4.02 m	83.4	2	4.03 m	75.0
14		45.9	3	4.26 m	75.5	3	3.68 m	88.0
15	1.82 m, 1.39 m	36.8	4	5.10 m	75.7	4	4.07 m	70.6
16	2.39 m, 2.97 m	35.6	5	3.70 m, 4.69 m	64.3	5	3.97 m	78.3
17		89.8	Qui			6	4.20 m, 4.46 m	62.1
18		174.5	1	5.02 d (7.8 Hz)	105.4	OMe	3.84 s	60.8
19	1.36 s	22.5	2	3.96 m	76.3			
20		86.7	3	4.08 m	74.0			

Tang, 2006).

Compound **3**: white crystal. It gave a positive reaction to Libermann-Burchard and Molish tests. mp 235–236 °C, $[\alpha]_{\text{D}}^{20}$ –11.5° (*c* 0.4, MeOH); IR $\nu_{\text{max}}^{\text{KBr}}$ (cm^{-1}): 3421, 1761, 1653, 1256, 1073; ESI-MS (+) mode: m/z 1227 $[\text{M} + \text{Na}]^+$ ($\text{C}_{54}\text{H}_{85}\text{Na}_2\text{O}_{26}\text{S}^+$), 1125 $[\text{M} - \text{SO}_3\text{Na} + \text{H} + \text{Na}]^+$, ESI-MS (–) mode: m/z 1181 $[\text{M} - \text{Na}]^-$. It was deduced as 24-dehydroechinoside A by comparison of the NMR spectra with those of 24-dehydroechinoside (Kitagawa, Kobayashi, and Kyogoku, 1982; Kobayashi *et al.*, 1991).

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