# A New Secoiridoid Glycoside from Roots of Picrorhiza scrophulariiflora

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Abstract: Objective To study the chemical constituents in the roots of *Picrorhiza scrophulariiflora*. Methods The chemical constituents in the roots of *P. scrophulariiflora* were separated and purified with chromatographic methods. The structures were elucidated by spectroscopic methods and chemical analyses. Results A new secoiridoid glycoside, picrogentioside II (1) was successfully isolated from the roots of *P. scrophulariiflora*. Conclusion Compound 1 is a new secoiridoid glycoside.

Key words: picrogentioside II; *Picrorhiza scrophulariiflora*; Scrophulariaceae; secoiridoid glycoside **DOI:** 10.3969/j.issn.1674-6348.2013.03.010

### Introduction

Picrorhiza scrophulariiflora Pennell (Scrophulariaceae) grows in the high altitude regions (over 4400 m) in the southeast of Tibet and the northwest of Yunnan in China. The roots of this plant are used in traditional Chinese medicine for the treatments of damp-heat dysentery, jaundice, and steaming of bone (Pharmacopeia Committee of P. R. China, 2010). In the previous articles, we reported the isolation of six new secoiridoid glycosides and five new phenylethanoid glycosides from the roots of P. scrophulariiflora (Zhu et al, 2008; Zou et al, 2010). This paper describes the structural elucidation and identification of a new secoiridoid glycoside, named picrogentioside II (1).

## Materials and methods

# **Plant materials**

The roots of *Picrorhiza scrophulariiflora* Pennell were collected in October 2006 in Sichuan province, China, and identified by Prof. SUN Qi-shi (Shenyang Pharmaceutical University). The voucher specimen (20060304) has been deposited in the Herbarium of

School of Traditional Chinese Materia Medica of Shenyang Pharmaceutical University, China.

# General

The UV spectra were recorded on a Milton Roy Spectronic 1201 Spectrophotometer, and the FTIR spectra were measured on a Perkin-Elmer 157G Infrared Spectrophotometer. The optical rotations were obtained using a Perkin-Elmer 241-MC Polarimeter. The <sup>1</sup>H-NMR (400 MHz) and <sup>13</sup>C-NMR (100 MHz) spectra were obtained on a Bruker AV400 Spectrometer with DMSO-d<sub>6</sub> as the solvent and TMS as an internal standard. The HR-ESI-MS data were measured with a Bruker AOEXIII 7.0 TESLA FTMS. HPLC was conducted using reverse phase columns (Mighty Sil RP-18 and 8, Kantho Chemical Co., Ltd.) with the MeOH-H<sub>2</sub>O solvent system. The column chromatography was carried out on silica gel (Qingdao Marine Chemical Company China, 200-300 meshes) and Sephadex LH-20 (Amersham Pharmacia Biotech AB). The silica GF254 for TLC was produced by Qingdao Marine Chemical Company, China and Merck Company.

### **Extraction and isolation**

The dried ground roots (underground parts) of P.

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scrophulariiflora (3.0 kg) were successively extracted for three times with 90% EtOH under reflux. After removal of the solvent in vacuo, the residue (1.6 kg) was suspended in H<sub>2</sub>O and then extracted successively with petroleum ether (bp 60—90 °C), EtOAc, and *n*-BuOH. The *n*-BuOH layer was concentrated *in vacuo* to yield a viscous residue (500 g), which was then dissolved in water (2 L) and subjected to macroporous D-101 resin column chromatography eluted successively with water and ethanol (water, 30%, 50%, and 100% ethanol). The 50% EtOH eluted fraction was evaporated *in vacuo* to yield a residue (140 g) that was subjected to silica gel column chromatography eluted with CHCl<sub>3</sub>-MeOH of increasing polarity to give eight fractions (Frs. 1—8).

Fr. 3 (28.5 g) was chromatographed over silica gel using an EtOAc-MeOH gradient system as the eluent to yield five fractions: Frs. 3A—3E. Fr. 3B (276.6 mg) was purified by Sephadex LH-20 chromatography and further separated by reversed-phase HPLC using MeOH-H<sub>2</sub>O (57:43 $\rightarrow$ 48:52) as the mobile phase to yield compound **1** (13.7 mg).

#### **Results and discussion**

Compound **1** was obtained as yellow amorphous powder,  $[\alpha]_D^{20}$  –236.5° (*c* 0.51, MeOH) and its molecular formula was determined as C<sub>31</sub>H<sub>36</sub>O<sub>17</sub> by the

 $[M + Na]^+$  quasi-ion peak at *m/z* 703.1861 (calcd. for C<sub>31</sub>H<sub>36</sub>O<sub>17</sub>Na 703.1849) in the HR-ESI-MS. UV (214 and 278 nm). And IR  $v_{max}^{KBr}$  (cm<sup>-1</sup>): 3437, 1703, 1629, 1607, 1513 absorptions suggested the existence of hydroxyl, α, β-unsaturated ester, and aromatic groups in compound **1**.

The <sup>1</sup>H-NMR spectrum of compound 1 (Table 1) showed the presence of an acetal proton [ $\delta_{\rm H}$  5.60 (1H, d, J = 3.2 Hz), H-1], an oxygenated methylene proton [ $\delta_{\rm H}$ 5.00 (2H, m), H<sub>2</sub>-7], two olefinic protons [ $\delta_{\rm H}$  7.40 (1H, brs), H-3; 5.65 (1H, m), H-6], one methine proton [ $\delta_{\rm H}$ 3.32, (1H, m), H-9], and one terminal vinyl group [ $\delta_{\rm H}$ 5.20 (1H, dd, J = 2.0, 10.4 Hz, H<sub>1</sub>-10, 5.22 (1H, dd, J =2.0, 16.8 Hz), H<sub>2</sub>-10 and 5.73 (1H, ddd, J = 6.8, 10.4, 16.8 Hz), H-8], due to one gentiopicroside moiety (Tian, Zhao, and Luan, 2007). Furthermore, the spectrum showed the presence of a trans-caffeoyl moiety (Zhu et al, 2008), which was evident from three aromatic H-atoms forming an ABX system [ $\delta$  (A) 7.15,  $\delta$  (B) 7.11,  $\delta$  (X) 7.11], and two H-atoms of an *trans*configured C=C bond at  $\delta$  (H) 6.40, 7.51 (2d, J = 16.0 Hz each).

In addition, two anomeric H-atoms which were observed at  $\delta_{\rm H}$  4.66 (1H, d, J = 8.0 Hz) and  $\delta_{\rm H}$  4.78 (1H, d, J = 6.8 Hz) and combined with the <sup>13</sup>C-NMR signals at  $\delta_{\rm C}$  101.6—60.7 (Table 1), are typical characters for

Positions	$\delta_{ m C}$	$\delta_{ m H}$	Positions	$\delta_{ m C}$	$\delta_{ m H}$
secoiridoid			7''	144.1	7.51, d, (16.0)
1	96.5	5.60, d, (3.2)	8″	116.6	6.40, d, (16.0)
3	148.6	7.40, brs	9″	165.9	
4	103.4		glucose-1		
5	124.9		1′	98.5	4.66, d, (8.0)
6	116.3	5.65, m	2'	70.9	3.18, m
7	69.1	5.00, m	3'	77.6	4.90, t, (9.2)
8	134.0	5.73, ddd, (6.8, 10.4, 16.8)	4'	67.8	3.34, m
9	44.4	3.32, m	5'	77.0	3.34, m
10	118.0	5.20, dd, (2.0, 10.4);	6'	60.7	3.47, m;
11	162.7		glucose-2		
caffeoyl			1‴	101.6	4.78, d, (6.8)
1″	128.7		2‴	75.8	3.34, m
2″	114.9	7.15, brs	3‴	73.2	3.31, m
3″	146.8		4‴	69.8	3.18, m
4″	147.3		5‴	77.2	3.34, m
5″	116.1	7.11, brs	6‴	60.7	3.47, m;
6"	120.7	7.11, brs			

Table 1 <sup>13</sup>C-NMR (100 MHz) and <sup>1</sup>H-NMR (400 MHz) data of compound 1 (DMSO-d<sub>6</sub>)

two  $\beta$ -glucopyranosyl moieties. The sugar component was identified as *D*-glucopyranoside by co-HPLC analysis of the standard sugar (Oshima, Yamauchi, and Kumanotani, 1982). The relatively large *J* value of the two anomeric protons indicated that the glucoside linkage had a  $\beta$ -configuration.

The detailed analysis on <sup>1</sup>H-NMR, TOCSY, <sup>13</sup>C-NMR, HMQC, and HMBC spectra of compound 1 fixed the connections between the various moieties. The HMBC (Fig. 1) correlation between H-1" ( $\delta_{\rm H}$  4.78) and C-4" ( $\delta_{\rm C}$  147.3) demonstrated that glucose 2 was connected to the C-4" oxygen atom. The downfield shift of the oxygenated methenyl proton at  $\delta_{\rm H}$  4.90 (H-3') suggested that the caffeoyl moiety was attached at C-3' of the glucose 1. This was confirmed by HMBC correlation between H-3' ( $\delta_{\rm H}$  4.90) and the carbonyl carbon C-9" ( $\delta_{\rm C}$  165.9). The relative stereochemistry of compound 1 was determined from the NOESY spectrum. The correlation between H-1 ( $\delta_{\rm H}$  5.60) and H-8 ( $\delta_{\rm H}$  5.73) showed that the two protons were positioned on the same side. Thus, the structure of compound 1 was identified as in Fig. 2.



Fig. 1 HMBC (H $\rightarrow$ C) and NOESY (\* ) correlations for picrogentioside II



Fig. 2 Structure of picrogentioside II

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